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Butyl 4-(4-methylbenzenesulfonamido)-benzoate

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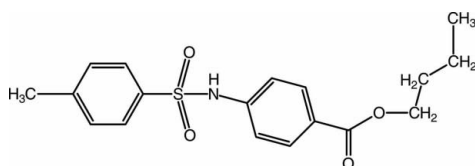
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 Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; disorder in main residue; R factor = 0.072; wR factor = 0.218; data-to-parameter ratio = 15.9.

In the title compound, $\text{C}_{18}\text{H}_{21}\text{NO}_4\text{S}$, the aromatic rings are almost normal to each other, with a dihedral angle of $89.27(18)^\circ$. The molecular conformation is stabilized by an intramolecular $\text{C}-\text{H}\cdots\text{O}$ interaction, which generates an $S(6)$ motif. In the crystal, $\text{N}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds lead to the formation of chains propagating along $[010]$. Neighbouring chains are linked *via* a $\text{C}-\text{H}\cdots\pi$ interaction. The $-\text{CH}_2\text{CH}_2\text{CH}_3$ atoms of the butyl group are disordered over two sets of sites, with a refined site-occupancy ratio of 0.536 (16):0.464 (16).

Related literature

For related structures, see: Mustafa *et al.* (2010, 2011, 2012); Khan *et al.* (2011). For bond-length data, see: Allen *et al.* (1987). For the graph-set analysis of hydrogen bonding, see: Bernstein *et al.* (1995).



Experimental

Crystal data

 $\text{C}_{18}\text{H}_{21}\text{NO}_4\text{S}$
 $M_r = 347.43$
 Monoclinic, $P2_1/c$
 $a = 17.8216(13)$ Å
 $b = 8.2702(6)$ Å

 $c = 11.9282(8)$ Å
 $\beta = 91.001(3)^\circ$
 $V = 1757.8(2)$ Å³
 $Z = 4$
 Mo $K\alpha$ radiation

 $\mu = 0.21$ mm⁻¹
 $T = 296$ K

 $0.33 \times 0.25 \times 0.21$ mm

Data collection

 Bruker APEXII CCD
 diffractometer
 13164 measured reflections

 3557 independent reflections
 2287 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.038$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.072$
 $wR(F^2) = 0.218$
 $S = 1.05$
 3557 reflections

 224 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.47$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.39$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C2–C7 benzene ring.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{O3}^{\text{i}}$	0.86	2.11	2.868 (4)	146
$\text{C9}-\text{H9}\cdots\text{O2}$	0.93	2.36	3.015 (4)	127
$\text{C10}-\text{H10}\cdots\text{O1}^{\text{ii}}$	0.93	2.53	3.453 (4)	173
$\text{C1}-\text{H1C}\cdots\text{Cg1}^{\text{iii}}$	0.96	2.76	3.639 (6)	153

 Symmetry codes: (i) $x, y - 1, z$; (ii) $x, y + 1, z$; (iii) $-x, -y, -z + 1$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SU2404).

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supplementary materials

Acta Cryst. (2012). E68, o1541 [doi:10.1107/S1600536812015413]

Butyl 4-(4-methylbenzenesulfonamido)benzoate**Ghulam Mustafa, Mehmet Akkurt, Yılmaz Dağdemir and Islam Ullah Khan****Comment**

As part of our ongoing studies of sulfonamides with potential biological properties (Mustafa *et al.*, 2010, 2011, 2012; Khan *et al.*, 2011), we describe herein the synthesis and crystal structure of the title compound.

As seen in Fig. 1, the two aromatic rings (C2—C7) and (C8—C13) are almost normal to each other, with a dihedral angle of 89.27 (18)°. The S atom has a distorted tetrahedral coordination geometry [S1—O1 = 1.411 (3), S1—O2 = 1.419 (3), S1—N1 = 1.626 (3), S1—C5 = 1.760 (4) Å, O1—S1—O2 = 120.43 (15), O1—S1—N1 = 105.23 (17), O1—S1—C5 = 106.92 (16), O2—S1—N1 = 109.21 (16), O2—S1—C5 = 108.23 (17) and N1—S1—C5 = 105.94 (15)°]. All the bond lengths (Allen *et al.*, 1987) and angles are within normal ranges and are comparable to those found for similar structures (Mustafa *et al.*, 2010, 2011, 2012; Khan *et al.*, 2011).

The molecular conformation of the title compound is stabilized by an intramolecular C—H···O interaction, generating an S(6) motif (Table 1; Bernstein *et al.*, 1995). In the crystal, N—H···O and C—H···O hydrogen bonds lead to the formation of chains propagating along [010] - see Fig. 2 and Table 1. Neighbouring chains are linked *via* a C—H··· π interaction (Table 1).

Experimental

To an aqueous solution of *p*-amino benzoic acid (1.0 g, 7.3 mmol), sodium carbonate (1 N) was added to adjust the pH to 8. *p*-toluenesulfonyl chloride (1.80 g, 9.48 mmol) was then added and the mixture was stirred at room temperature keeping the pH of the mixture up to 8 with occasional addition of sodium carbonate solution. The progress and completion of the reaction was confirmed by TLC and conversion of the suspension into a clear solution. After 2 h, the mixture was poured into a beaker and the pH was adjusted to 2.0 by addition of 1 N HCl. Precipitates were produced which were filtered and washed with distilled water. The prepared sulfonamide (4-(toluene-4-sulfonylamino)-benzoic acid) (1.0 g, 3.43 mmol), DMF (10 ml) and *n*-hexane washed with sodium hydride (0.25 g, 10.31 mmol) were stirred at room temperature for 40 min, followed by the addition of butyl iodide (0.94 g, 5.15 mmol). The whole reaction mixture was stirred till the completion of the reaction and poured into crushed ice in a beaker. The pH of the mixture was adjusted to 4.0 with 1 N HCl. Precipitates were produced, filtered and washed twice with distilled water. Crystallization in chloroform gave long block-like pale-yellow X-ray quality crystals of the title compound.

Refinement

All the H-atoms were included in calculated positions and treated as riding atoms: N—H = 0.88 (2) Å, C—H = 0.93, 0.96 and 0.97 Å for CH, CH₃ and CH₂ H-atoms, respectively, with $U_{iso}(H) = k \times U_{eq}(N,C)$, where $k = 1.5$ for CH₃ H-atoms and $= 1.2$ for all other H-atoms. The —CH₂—CH₂—CH₃ atoms (C16, C17 and C18) of the butyl group are disordered over two sets of sites (A/B), with a refined site occupancy ratio of 0.536 (16):0.464 (16). Twelve poorly fitted reflections (1 0 0), (1 1 2), (−1 4 2), (1 3 4), (0 2 2), (−11 3 6), (8 3 0), (6 1 2), (2 3 8), (11 1 0), (−7 2 2) and (−14 3 4) were omitted from the

refinement.

Computing details

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT* (Bruker, 2007); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON* (Spek, 2009).

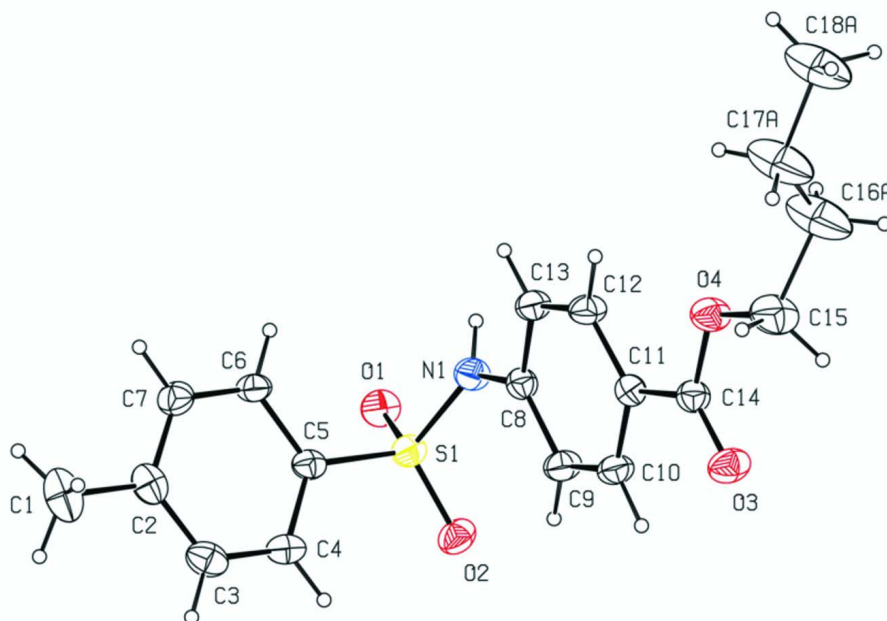
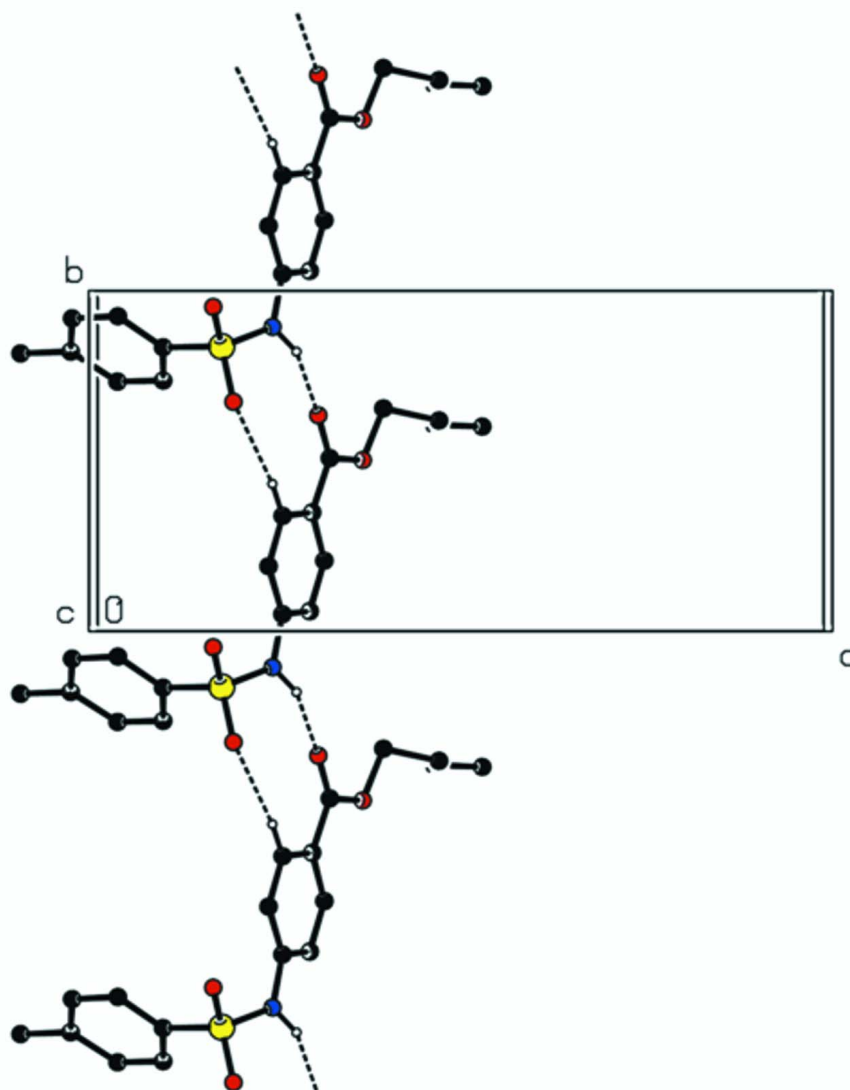


Figure 1

The molecular structure of the title molecule, with the atom numbering. Displacement ellipsoids are drawn at the 20% probability level. Only the atoms of the major disordered component of the butyl group are shown.


Figure 2

A partial view along the *c* axis of the crystal packing of the title compound. The N—H···O and C—H···O hydrogen bonds are shown as dashed lines (see Table 1 for details). Only the atoms of the major disordered component of the terminal butyl group are shown.

Butyl 4-(4-methylbenzenesulfonamido)benzoate

Crystal data

$C_{18}H_{21}NO_4S$

$M_r = 347.43$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1/c$

$a = 17.8216$ (13) Å

$b = 8.2702$ (6) Å

$c = 11.9282$ (8) Å

$\beta = 91.001$ (3)°

$V = 1757.8$ (2) Å³

$Z = 4$

$F(000) = 736$

$D_x = 1.313$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 2942 reflections

$\theta = 2.7$ – 21.5 °

$\mu = 0.21$ mm⁻¹

$T = 296$ K
Long block, light yellow

$0.33 \times 0.25 \times 0.21$ mm

Data collection

Bruker APEXII CCD
diffractometer
Radiation source: sealed tube
Graphite monochromator
 φ and ω scans
13164 measured reflections
3557 independent reflections

2287 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.038$
 $\theta_{\text{max}} = 26.4^\circ$, $\theta_{\text{min}} = 3.0^\circ$
 $h = -21 \rightarrow 22$
 $k = -10 \rightarrow 10$
 $l = -11 \rightarrow 14$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.072$
 $wR(F^2) = 0.218$
 $S = 1.05$
3557 reflections
224 parameters
0 restraints
Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map
Hydrogen site location: inferred from
neighbouring sites
H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.1157P)^2 + 0.8149P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.47 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.39 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.17821 (5)	-0.16383 (11)	0.79992 (7)	0.0561 (3)	
O1	0.19456 (16)	-0.3261 (3)	0.8276 (2)	0.0672 (9)	
O2	0.16764 (16)	-0.0474 (3)	0.88556 (19)	0.0707 (10)	
O3	0.30722 (17)	0.6342 (3)	0.5914 (2)	0.0779 (10)	
O4	0.36618 (16)	0.5025 (3)	0.4580 (2)	0.0732 (10)	
N1	0.24647 (17)	-0.1059 (4)	0.7206 (2)	0.0618 (10)	
C1	-0.0991 (3)	-0.1840 (7)	0.5042 (4)	0.0992 (19)	
C2	-0.0300 (2)	-0.1785 (5)	0.5775 (3)	0.0666 (14)	
C3	-0.0269 (2)	-0.0810 (5)	0.6726 (4)	0.0725 (17)	
C4	0.0361 (2)	-0.0747 (5)	0.7391 (3)	0.0618 (12)	
C5	0.09731 (19)	-0.1661 (4)	0.7132 (3)	0.0506 (10)	
C6	0.0964 (2)	-0.2636 (4)	0.6183 (3)	0.0582 (12)	
C7	0.0334 (2)	-0.2668 (5)	0.5514 (3)	0.0659 (16)	
C8	0.25941 (18)	0.0499 (4)	0.6752 (3)	0.0515 (11)	
C9	0.2412 (2)	0.1907 (4)	0.7308 (3)	0.0641 (14)	
C10	0.2598 (2)	0.3388 (4)	0.6868 (3)	0.0625 (12)	

C11	0.29879 (18)	0.3488 (4)	0.5867 (3)	0.0528 (11)	
C12	0.3148 (2)	0.2070 (5)	0.5310 (3)	0.0618 (12)	
C13	0.2953 (2)	0.0585 (4)	0.5737 (3)	0.0582 (12)	
C14	0.3224 (2)	0.5084 (4)	0.5476 (3)	0.0573 (12)	
C15	0.3938 (3)	0.6546 (6)	0.4172 (5)	0.102 (2)	
C16A	0.4688 (9)	0.619 (2)	0.3570 (12)	0.152 (5)	0.536 (16)
C17A	0.4608 (8)	0.607 (2)	0.2481 (12)	0.152 (5)	0.536 (16)
C18A	0.5254 (9)	0.601 (2)	0.1612 (13)	0.152 (5)	0.536 (16)
C17B	0.5053 (7)	0.5832 (16)	0.3103 (10)	0.086 (3)	0.464 (16)
C18B	0.5446 (7)	0.5739 (17)	0.2028 (10)	0.086 (3)	0.464 (16)
C16B	0.4245 (7)	0.6307 (15)	0.2980 (10)	0.086 (3)	0.464 (16)
H1B	-0.10320	-0.28850	0.46960	0.1490*	
H3	-0.06840	-0.01930	0.69120	0.0870*	
H4	0.03750	-0.00810	0.80190	0.0740*	
H1C	-0.09590	-0.10240	0.44720	0.1490*	
H1	0.27880	-0.17900	0.70400	0.0740*	
H1A	-0.14250	-0.16450	0.54880	0.1490*	
H10	0.24620	0.43290	0.72400	0.0750*	
H12	0.33930	0.21190	0.46280	0.0740*	
H13	0.30630	-0.03550	0.53440	0.0700*	
H15B	0.40420	0.72770	0.47910	0.1220*	0.536 (16)
H15D	0.35690	0.70440	0.36740	0.1220*	0.536 (16)
H16C	0.50410	0.70580	0.37400	0.1820*	0.536 (16)
H16D	0.49000	0.51960	0.38630	0.1820*	0.536 (16)
H17C	0.42890	0.69650	0.22530	0.1820*	0.536 (16)
H17D	0.43160	0.50960	0.23530	0.1820*	0.536 (16)
H18D	0.57300	0.60340	0.20010	0.2280*	0.536 (16)
H18E	0.52150	0.69330	0.11240	0.2280*	0.536 (16)
H18F	0.52130	0.50390	0.11780	0.2280*	0.536 (16)
H6	0.13810	-0.32570	0.60060	0.0700*	
H7	0.03300	-0.32990	0.48680	0.0790*	
H9	0.21610	0.18520	0.79840	0.0770*	
H15A	0.43340	0.69440	0.46680	0.1220*	0.464 (16)
H15C	0.35360	0.73360	0.41540	0.1220*	0.464 (16)
H16A	0.39650	0.54670	0.25890	0.1030*	0.464 (16)
H16B	0.42000	0.73020	0.25530	0.1030*	0.464 (16)
H17A	0.53080	0.66120	0.35830	0.1030*	0.464 (16)
H17B	0.50830	0.47870	0.34710	0.1030*	0.464 (16)
H18A	0.50870	0.55730	0.14300	0.1280*	0.464 (16)
H18B	0.57940	0.48530	0.20490	0.1280*	0.464 (16)
H18C	0.57130	0.67290	0.19040	0.1280*	0.464 (16)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0664 (6)	0.0498 (5)	0.0523 (5)	-0.0032 (4)	0.0105 (4)	0.0025 (4)
O1	0.0839 (17)	0.0506 (15)	0.0674 (15)	-0.0016 (13)	0.0124 (13)	0.0131 (12)
O2	0.0923 (19)	0.0690 (17)	0.0512 (13)	-0.0142 (14)	0.0126 (13)	-0.0087 (12)
O3	0.102 (2)	0.0505 (16)	0.0821 (18)	-0.0024 (14)	0.0267 (16)	-0.0034 (14)

O4	0.0831 (18)	0.0614 (17)	0.0760 (17)	-0.0054 (13)	0.0292 (15)	0.0018 (13)
N1	0.0649 (18)	0.0483 (16)	0.0727 (19)	-0.0003 (14)	0.0159 (15)	0.0005 (14)
C1	0.079 (3)	0.122 (4)	0.096 (3)	-0.023 (3)	-0.016 (3)	0.043 (3)
C2	0.065 (2)	0.071 (3)	0.064 (2)	-0.010 (2)	0.0044 (18)	0.026 (2)
C3	0.069 (3)	0.069 (3)	0.080 (3)	0.014 (2)	0.014 (2)	0.013 (2)
C4	0.073 (2)	0.056 (2)	0.057 (2)	0.0118 (18)	0.0151 (18)	-0.0002 (17)
C5	0.0594 (19)	0.0450 (18)	0.0479 (17)	-0.0002 (15)	0.0139 (15)	0.0061 (14)
C6	0.061 (2)	0.060 (2)	0.054 (2)	0.0045 (17)	0.0137 (17)	-0.0102 (16)
C7	0.075 (3)	0.073 (3)	0.050 (2)	-0.005 (2)	0.0067 (19)	-0.0060 (18)
C8	0.0508 (18)	0.0503 (19)	0.0536 (18)	-0.0025 (15)	0.0047 (15)	0.0010 (15)
C9	0.078 (3)	0.054 (2)	0.061 (2)	-0.0025 (18)	0.0210 (19)	-0.0067 (17)
C10	0.070 (2)	0.052 (2)	0.066 (2)	0.0026 (17)	0.0158 (18)	-0.0109 (17)
C11	0.0501 (18)	0.054 (2)	0.0545 (18)	-0.0032 (15)	0.0042 (15)	-0.0006 (16)
C12	0.071 (2)	0.061 (2)	0.054 (2)	0.0027 (18)	0.0191 (17)	-0.0031 (17)
C13	0.066 (2)	0.050 (2)	0.059 (2)	-0.0004 (17)	0.0144 (17)	-0.0052 (16)
C14	0.057 (2)	0.054 (2)	0.061 (2)	0.0013 (16)	0.0057 (17)	-0.0010 (17)
C15	0.118 (4)	0.074 (3)	0.115 (4)	-0.012 (3)	0.056 (3)	0.015 (3)
C16A	0.107 (7)	0.237 (12)	0.113 (7)	0.056 (6)	0.042 (5)	0.048 (6)
C17A	0.107 (7)	0.237 (12)	0.113 (7)	0.056 (6)	0.042 (5)	0.048 (6)
C18A	0.107 (7)	0.237 (12)	0.113 (7)	0.056 (6)	0.042 (5)	0.048 (6)
C17B	0.062 (5)	0.104 (5)	0.091 (6)	-0.005 (3)	0.007 (3)	0.015 (4)
C18B	0.062 (5)	0.104 (5)	0.091 (6)	-0.005 (3)	0.007 (3)	0.015 (4)
C16B	0.062 (5)	0.104 (5)	0.091 (6)	-0.005 (3)	0.007 (3)	0.015 (4)

Geometric parameters (Å, °)

S1—O1	1.411 (3)	C1—H1A	0.9600
S1—O2	1.419 (3)	C1—H1B	0.9600
S1—N1	1.626 (3)	C1—H1C	0.9600
S1—C5	1.760 (4)	C3—H3	0.9300
O3—C14	1.197 (4)	C4—H4	0.9300
O4—C14	1.335 (4)	C6—H6	0.9300
O4—C15	1.439 (6)	C7—H7	0.9300
N1—C8	1.418 (5)	C9—H9	0.9300
N1—H1	0.8600	C10—H10	0.9300
C1—C2	1.499 (6)	C12—H12	0.9300
C2—C3	1.392 (6)	C13—H13	0.9300
C2—C7	1.385 (5)	C15—H15B	0.9700
C3—C4	1.364 (5)	C15—H15D	0.9700
C4—C5	1.367 (5)	C15—H15A	0.9700
C5—C6	1.390 (5)	C15—H15C	0.9700
C6—C7	1.366 (5)	C16A—H16C	0.9700
C8—C13	1.381 (5)	C16A—H16D	0.9700
C8—C9	1.382 (5)	C16B—H16B	0.9700
C9—C10	1.375 (5)	C16B—H16A	0.9700
C10—C11	1.394 (5)	C17A—H17D	0.9700
C11—C12	1.380 (5)	C17A—H17C	0.9700
C11—C14	1.464 (5)	C17B—H17A	0.9700
C12—C13	1.377 (5)	C17B—H17B	0.9700
C15—C16B	1.545 (13)	C18A—H18E	0.9600

C15—C16A	1.556 (17)	C18A—H18F	0.9600
C16A—C17A	1.31 (2)	C18A—H18D	0.9600
C16B—C17B	1.497 (18)	C18B—H18A	0.9600
C17A—C18A	1.56 (2)	C18B—H18B	0.9600
C17B—C18B	1.474 (17)	C18B—H18C	0.9600
O1—S1—O2	120.43 (15)	C6—C7—H7	119.00
O1—S1—N1	105.23 (17)	C8—C9—H9	120.00
O1—S1—C5	106.92 (16)	C10—C9—H9	120.00
O2—S1—N1	109.21 (16)	C9—C10—H10	120.00
O2—S1—C5	108.23 (17)	C11—C10—H10	120.00
N1—S1—C5	105.94 (15)	C11—C12—H12	119.00
C14—O4—C15	116.6 (3)	C13—C12—H12	119.00
S1—N1—C8	128.2 (3)	C8—C13—H13	120.00
C8—N1—H1	116.00	C12—C13—H13	120.00
S1—N1—H1	116.00	O4—C15—H15B	110.00
C3—C2—C7	117.9 (3)	O4—C15—H15D	110.00
C1—C2—C7	121.2 (4)	O4—C15—H15A	110.00
C1—C2—C3	120.9 (4)	O4—C15—H15C	110.00
C2—C3—C4	121.1 (4)	C16A—C15—H15B	108.00
C3—C4—C5	120.0 (4)	C16A—C15—H15D	112.00
S1—C5—C6	118.9 (3)	H15B—C15—H15D	109.00
S1—C5—C4	120.7 (3)	C16B—C15—H15A	110.00
C4—C5—C6	120.5 (3)	C16B—C15—H15C	110.00
C5—C6—C7	119.0 (3)	H15A—C15—H15C	108.00
C2—C7—C6	121.6 (3)	C15—C16A—H16C	109.00
N1—C8—C9	122.8 (3)	C15—C16A—H16D	109.00
N1—C8—C13	117.6 (3)	C17A—C16A—H16C	109.00
C9—C8—C13	119.6 (3)	C17A—C16A—H16D	109.00
C8—C9—C10	120.5 (3)	H16C—C16A—H16D	108.00
C9—C10—C11	120.4 (3)	H16A—C16B—H16B	109.00
C12—C11—C14	123.4 (3)	C15—C16B—H16A	110.00
C10—C11—C14	118.5 (3)	C15—C16B—H16B	110.00
C10—C11—C12	118.2 (3)	C17B—C16B—H16A	110.00
C11—C12—C13	121.6 (3)	C17B—C16B—H16B	110.00
C8—C13—C12	119.6 (3)	C18A—C17A—H17C	106.00
O3—C14—C11	125.2 (3)	C18A—C17A—H17D	106.00
O3—C14—O4	121.4 (3)	C16A—C17A—H17D	106.00
O4—C14—C11	113.4 (3)	C16A—C17A—H17C	106.00
O4—C15—C16B	109.2 (6)	H17C—C17A—H17D	106.00
O4—C15—C16A	107.0 (7)	C16B—C17B—H17A	109.00
C15—C16A—C17A	113.1 (12)	C16B—C17B—H17B	109.00
C15—C16B—C17B	107.4 (9)	H17A—C17B—H17B	108.00
C16A—C17A—C18A	126.3 (14)	C18B—C17B—H17B	109.00
C16B—C17B—C18B	113.5 (10)	C18B—C17B—H17A	109.00
C2—C1—H1A	109.00	C17A—C18A—H18F	110.00
C2—C1—H1B	109.00	C17A—C18A—H18E	109.00
C2—C1—H1C	109.00	H18E—C18A—H18F	110.00
H1A—C1—H1B	109.00	H18D—C18A—H18E	109.00

H1A—C1—H1C	109.00	H18D—C18A—H18F	110.00
H1B—C1—H1C	109.00	C17A—C18A—H18D	109.00
C2—C3—H3	120.00	C17B—C18B—H18A	109.00
C4—C3—H3	119.00	C17B—C18B—H18B	109.00
C3—C4—H4	120.00	C17B—C18B—H18C	110.00
C5—C4—H4	120.00	H18A—C18B—H18B	109.00
C5—C6—H6	120.00	H18A—C18B—H18C	109.00
C7—C6—H6	121.00	H18B—C18B—H18C	109.00
C2—C7—H7	119.00		
O1—S1—N1—C8	-176.1 (3)	C3—C4—C5—C6	1.1 (6)
O2—S1—N1—C8	-45.5 (3)	S1—C5—C6—C7	179.4 (3)
C5—S1—N1—C8	70.9 (3)	C4—C5—C6—C7	-0.1 (5)
O1—S1—C5—C4	126.3 (3)	C5—C6—C7—C2	-1.6 (6)
O1—S1—C5—C6	-53.2 (3)	N1—C8—C9—C10	175.3 (3)
O2—S1—C5—C4	-4.8 (4)	C13—C8—C9—C10	-1.2 (5)
O2—S1—C5—C6	175.7 (3)	N1—C8—C13—C12	-174.5 (3)
N1—S1—C5—C4	-121.9 (3)	C9—C8—C13—C12	2.2 (5)
N1—S1—C5—C6	58.7 (3)	C8—C9—C10—C11	-1.4 (5)
C15—O4—C14—O3	-0.1 (5)	C9—C10—C11—C12	3.0 (5)
C15—O4—C14—C11	-177.9 (3)	C9—C10—C11—C14	-175.4 (3)
C14—O4—C15—C16A	153.3 (6)	C10—C11—C12—C13	-2.0 (5)
S1—N1—C8—C9	33.7 (5)	C14—C11—C12—C13	176.3 (3)
S1—N1—C8—C13	-149.7 (3)	C10—C11—C14—O3	-4.6 (5)
C1—C2—C3—C4	-179.5 (4)	C10—C11—C14—O4	173.1 (3)
C7—C2—C3—C4	-1.1 (6)	C12—C11—C14—O3	177.1 (4)
C1—C2—C7—C6	-179.4 (4)	C12—C11—C14—O4	-5.2 (5)
C3—C2—C7—C6	2.2 (6)	C11—C12—C13—C8	-0.6 (5)
C2—C3—C4—C5	-0.5 (6)	O4—C15—C16A—C17A	98.0 (13)
C3—C4—C5—S1	-178.3 (3)	C15—C16A—C17A—C18A	169.1 (13)

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1 is the centroid of the C2–C7 benzene ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 \cdots O3 ⁱ	0.86	2.11	2.868 (4)	146
C4—H4 \cdots O2	0.93	2.53	2.908 (4)	105
C9—H9 \cdots O2	0.93	2.36	3.015 (4)	127
C10—H10 \cdots O1 ⁱⁱ	0.93	2.53	3.453 (4)	173
C1—H1C \cdots Cg1 ⁱⁱⁱ	0.96	2.76	3.639 (6)	153

Symmetry codes: (i) $x, y-1, z$; (ii) $x, y+1, z$; (iii) $-x, -y, -z+1$.